

2,3-Dihydro-2-oxo-1*H*-benzimidazole-1-carboxamides with Selective Affinity for the 5-HT₄ Receptor: Synthesis and Structure–Affinity and Structure–Activity Relationships of a New Series of Partial Agonist and Antagonist Derivatives

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A series of 2,3-dihydro-2-oxo-1*H*-benzimidazole-1-carboxamide derivatives bearing a piperazine moiety was synthesized. Their *in vitro* 5-HT₄, 5-HT₃, and D₂ receptors affinities were evaluated by radioligand binding assay. For selected compounds functional studies at the 5-HT₄ receptor were made by using precontracted (by carbachol) preparations of rat esophageal tunica muscularis mucosae (TMM). The influence of the 3-substituent of the benzimidazole ring, the 4-substituent of the piperazine moiety, and the alkylene spacer was studied. Compounds with an ethyl or a cyclopropyl substituent in the 3-position of the benzimidazole ring showed moderate to high affinity ($K_i = 6.7\text{--}75.4$ nM) for the 5-HT₄ receptor with selectivity over 5-HT₃ and D₂ receptors and moderate antagonist activity ($pK_b = 6.19\text{--}7.73$). Compounds with an isopropyl substituent in the 3-benzimidazole position exhibited moderate and selective 5-HT₄ affinity ($K_i \geq 38.9$ nM) and a partial agonist activity (**5a**, i.a. = 0.94) higher than that of the reference compound BIMU 8 (i.a. = 0.70). This reversal of the pharmacological activity due only to a small structural difference might confirm the existence of two binding sites on the 5-HT₄ receptor. In the alkylene spacer, a two-methylene chain is favorable to optimize the affinity and the antagonist or the partial agonist activity. In the ethyl and cyclopropyl series, 5-HT₄ antagonist activity seems to be unrelated to the size of the 4-substituent of the piperazine moiety, whereas a methyl group is optimal for high partial agonist activity in the isopropyl series; however, the presence of a butyl substituent is a favorable pattern for 5-HT₄ antagonism and even causes a reversal of the pharmacological profile in the isopropyl series (**5h**, $pK_b = 7.94$). *N*-Butyl quaternization of **5a** led to an improvement in affinity for the 5-HT₄ receptor and maintained the high partial agonist activity (**5r**, $K_i = 66.3$ nM, i.a. = 0.93).

Introduction

The 5-HT₄ receptor was discovered in 1987, and shortly later it was shown that a series of gastrointestinal prokinetic benzamide derivatives (metoclopramide, renzapride, cisapride, etc.) act as agonists at the 5-HT₄ receptor.^{1–3} Since then, the 5-HT₄ receptor has been located in the central nervous system (CNS) (limbic system, hippocampo–habenulo–interpeduncular pathway, striato–nigro–tectal pathway) and peripheral tissues (gastrointestinal tract, heart, adrenal gland, urinary bladder).^{4–7} Considerable progress has been made with the development of specific radioligands, such as [³H]GR 113808 and [¹²⁵I]SB 207710.^{8,9} The 5-HT₄ receptor was cloned from rat brain in 1995, and it was found that it exists in two isoforms (5-HT_{4S} and 5-HT_{4L}) that differ in the length and sequence of their carboxy termini.^{10,11} Recently, the molecular structure and functional characterization of four splice variants of the human 5-HT₄ receptor have been described.¹²

To date, several 5-HT₄ receptor antagonists and agonists have been discovered.^{13,14} The chemical structures of the 5-HT₄ antagonists belong to five different classes including indolecarboxylates, benzoates, aryl ketones, imidazolopyridines, and benzimidazolones. On the other hand, there are six distinct classes of 5-HT₄ agonists including indoles, benzamides, benzoates, aryl

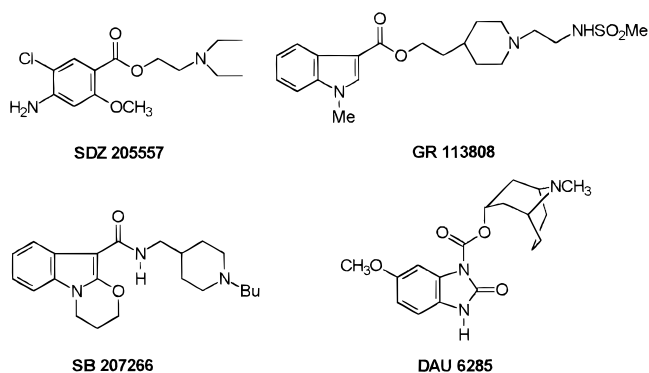


Figure 1. 5-HT₄ receptor antagonists.

ketones, 8-naphthalimides, and benzimidazolones. Some of the most known 5-HT₄ ligands are included in Figures 1 and 2.

Recently, three-dimensional maps of the 5-HT₄ agonist and antagonist recognition sites have been proposed.^{14–16} The structural features that define the 5-HT₄ pharmacophore can be regarded as an aromatic moiety, a coplanar carbonyl function, and a basic nitrogen atom; distances between pharmacophoric elements and their spatial orientations have been determined. For both 5-HT₄ receptor agonist and antagonist recognition sites, the existence of a hydrophobic pocket which would

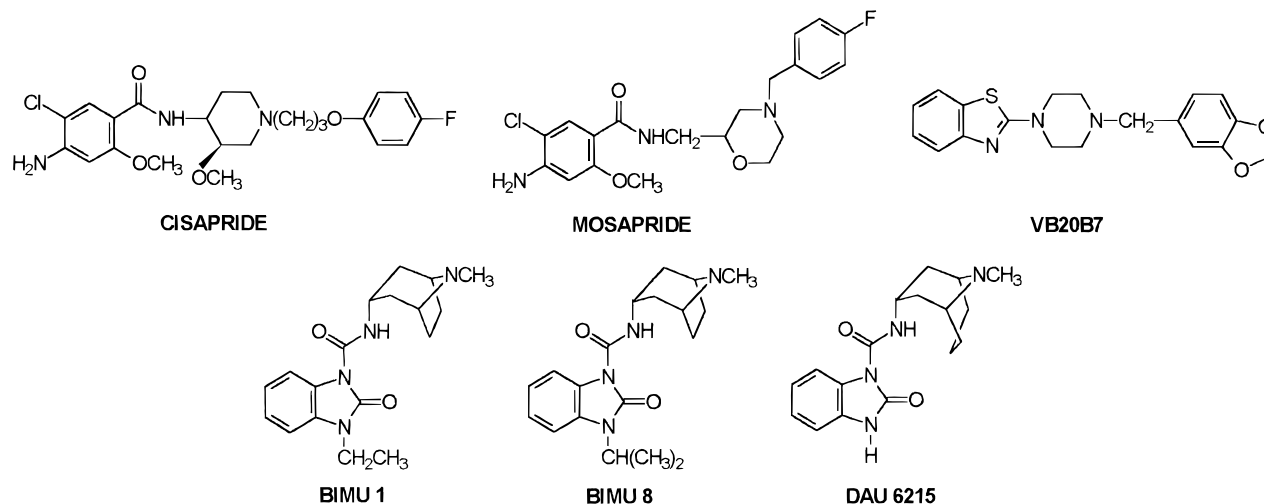


Figure 2. 5-HT₄ receptor agonists and DAU 6215 (5-HT₃ receptor antagonist).

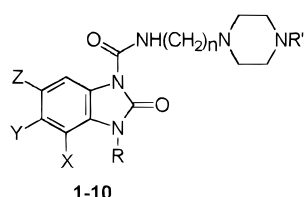


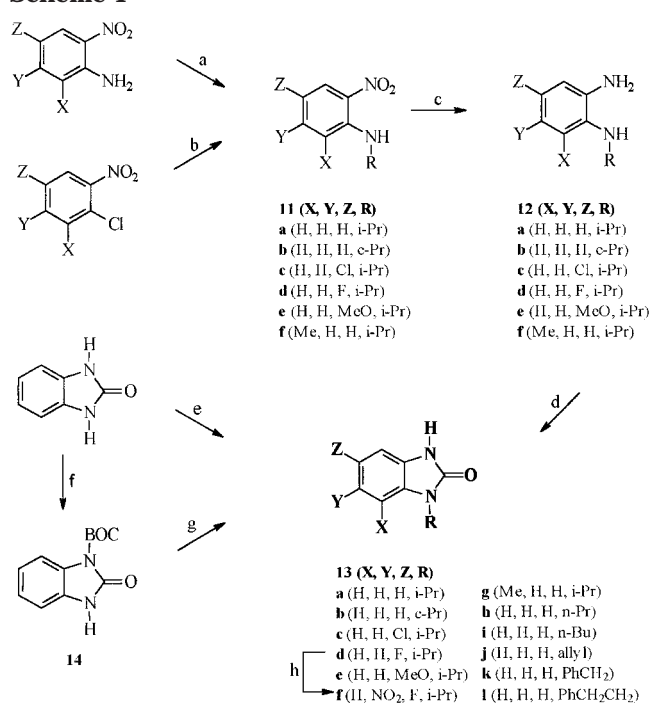
Figure 3. 2,3-Dihydro-2-oxo-1H-benzimidazole-1-carboxamides 1–10.

interact with the voluminous substituents of the basic nitrogen of active ligands is accepted. On the other hand, the existence of two binding sites at the 5-HT₄ receptor has been suggested to explain the agonist and antagonist activities of compounds with structural similarity.¹⁷

With regard to benzimidazolone derivatives, it has been reported that members of this class of compounds interact with both 5-HT₃ and 5-HT₄ receptors with different profiles of agonist or antagonist activity and are devoid of affinity for the dopamine receptors.^{18,19} Among them, DAU 6215 (Figure 2) is a selective 5-HT₃ receptor antagonist, whereas affinities of DAU 6285 (Figure 1) for 5-HT₃ and 5-HT₄ receptors are of the same order of magnitude and it behaves as a competitive 5-HT₄ antagonist.²⁰ In addition, BIMU 1 and BIMU 8 (Figure 2) combine 5-HT₃ antagonism and 5-HT₄ agonism.^{21–23}

Although the 5-HT₄ agonism or antagonism of some benzimidazolones is well-known, very little has been published about their 5-HT₄ receptor affinity and their structure–affinity relationships.²⁴ It is possible that the combination of 5-HT₃ antagonism and 5-HT₄ agonism will have advantages over a selective 5-HT₄ agonist action in some areas.²⁵ However, selective 5-HT₄ receptor ligands in this series could provide a useful tool to investigate the pharmacology of the 5-HT₄ receptor and to develop new therapies for the treatment of CNS disorders (cognition, emotional and motor functional control) and other peripheral disorders (irritable bowel syndrome, arrhythmia, urinary disturbance). Our aim was to find new benzimidazolones with affinity for the 5-HT₄ receptor and selectivity versus 5-HT₃ and D₂ receptors. In the present paper, we report the synthesis and the binding profile at 5-HT₄, 5-HT₃, and D₂ receptors of a series of 2,3-dihydro-2-oxo-1H-benzimidazole-1-carbox-

Scheme 1^a



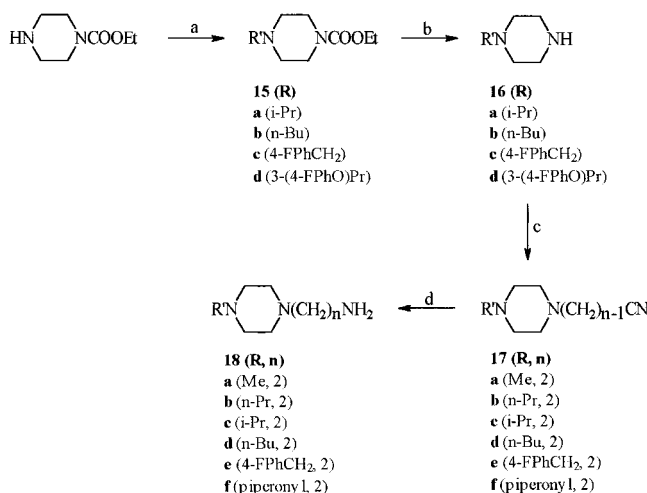
^a Reagents: (a) i, 2,2-dimethoxypropane, TFA, ii, H₃B–pyridine; (b) cyclopropylamine; (c) H₂, 10% Pd/C, or Zn, 1 N NaOH, EtOH; (d) CDI, THF; (e) HNa, DMF, RI; (f) HNa, DMF, (tBOC)₂O; (g) RBr, K₂CO₃, DMF; (h) HNO₃, Ac₂O.

amides including a piperazine moiety, 1–10 (Figure 3). We have studied the influence on the 5-HT₄ receptor affinity of each of these elements: the 3-substituent of the benzimidazolone ring, the 4-substituent of the piperazine moiety, and the alkylene spacer. In addition the functional activity at 5-HT₄ receptors has been also evaluated. Preliminary results in structure–affinity and structure–activity relationships are presented.

Chemistry

2,3-Dihydro-2-oxo-1H-benzimidazole-1-carboxamides 1–10 were prepared by using 2,3-dihydro-2-oxo-1H-benzimidazoles 13 and appropriate aminoalkylpiperazines 18 as key intermediates (Schemes 1–3).

According to Scheme 1 reductive amination of commercially available 2-nitroanilines by using 2,2-dimeth-

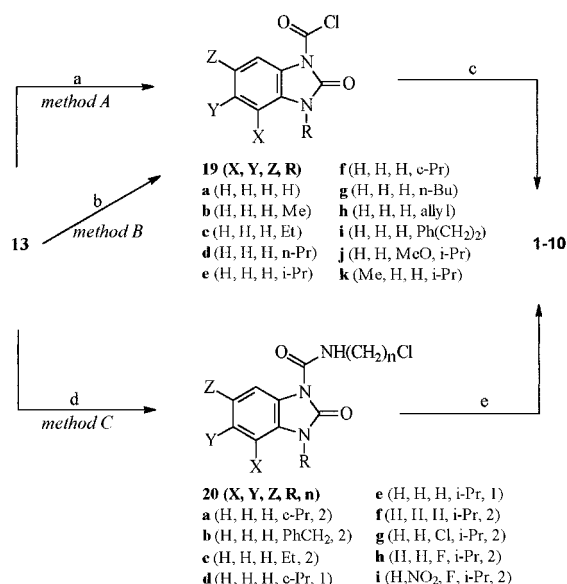
Scheme 2^a

^a Reagents: (a) K₂CO₃, CH₃CN, R'Br; (b) 48% HBr; (c) K₂CO₃, CH₃CN, Cl(CH₂)_{*n*-1}CN; (d) H₄LiAl, Et₂O.

oxypropane/trifluoroacetic acid and a boron–pyridine complex yielded the isopropyl derivatives **11** (R = *i*-Pr). Cyclopropyl derivatives **11** (R = *c*-Pr) were prepared following a previously described route.²⁶ 2-chloroni-trobenzene was heated with cyclopropylamine in a closed vessel to afford *N*-cyclopropyl-2-nitroaniline **11b** (X, Y, Z = H; R = *c*-Pr). Compounds **11** (R = *i*-Pr, *c*-Pr) were hydrogenated over palladium/carbon to give *o*-phenylenediamines **12** (R = *i*-Pr, *c*-Pr) which, by cyclization with *N,N*-carbonyldiimidazole, gave 1-alkyl-2,3-dihydro-2-oxo-1*H*-benzimidazoles **13** (R = *i*-Pr, *c*-Pr). Alkylation of commercially available 2,3-dihydro-2-oxo-1*H*-benzimidazole with sodium hydride and the corresponding alkyl halide provided a mixture of mono- and dialkylated derivatives; isolation of monoalkylated derivatives **13** (R = *n*-Pr, *n*-Bu, allyl, PhCH₂) was accomplished by flash chromatography. Treatment of 2,3-dihydro-2-oxo-1*H*-benzimidazole with di-*tert*-butyl dicarbonate according to reported conditions²⁷ yielded the alkoxycarbonyl-protected derivatives **14**, which were alkylated with an alkyl halide; the spontaneous elimination of the protecting group BOC provided derivatives **13** (R = PhCH₂CH₂).

Aminoalkylpiperazines **18** were prepared following standard procedures²⁸ (Scheme 2). Alkylation of *N*-carboxyethylpiperazine with a bromoalkane in acetonitrile in the presence of potassium carbonate afforded the corresponding derivatives **15**, which, after hydrolysis with 48% HBr, yielded the alkylpiperazines **16**; by alkylation with chloroalkylnitriles in acetonitrile and potassium carbonate, compounds **16** were converted to **17**, which, after reduction with lithium aluminum hydride in ether, provided the amines **18**.

2,3-Dihydro-2-oxo-1*H*-benzimidazole-1-carboxamides **1–10** were prepared through different routes of synthesis (Scheme 3). Chlorocarbonyl derivatives **19** were obtained according to previously reported conditions by employing trichloromethyl chloroformate in the presence of activated charcoal¹⁸ (method A) or by reaction with sodium hydride and phosgene²⁹ (method B). Coupling with substituted piperazines **18** yielded 2,3-dihydro-2-oxo-1*H*-benzimidazole-1-carboxamides **1–10**. Reaction of 2,3-dihydro-2-oxo-1*H*-benzimidazoles **13** with chloroalkyl isocyanates³⁰ provided chloroalkyl

Scheme 3^a

^a Reagents: (a) ClCO₂CCl₃, THF, charcoal; (b) 60% HNa, THF, phosgene; (c) **18**, TEA, THF; (d) Cl(CH₂)_{*n*}NCO, toluene; (e) K₂CO₃, toluene, KI, **16**, or TEA, CH₃CN, KI, **16**.

derivatives **20** (method C), which were converted into the 2,3-dihydro-2-oxo-1*H*-benzimidazole-1-carboxamides **1–10** by reaction with substituted piperazines **16**.

Pharmacology

The new benzimidazolone derivatives were initially evaluated for in vitro 5-HT₄ receptor affinity by radioligand binding assay. For each compound the ability to displace the specific ligand [³H]GR 113808 from 5-HT₄ receptors of guinea pig striatum was determined.³¹ Concentrations required to inhibit 50% of radioligand specific binding (IC₅₀) and K_i values were calculated from two separate competition experiments with samples in triplicate, using 10–12 different concentrations of displacer. Compounds with moderate to high affinity were selected for functional studies at the 5-HT₄ receptor by using rat isolated thoracic esophageal tunica muscularis mucosae (TMM) precontracted by carbachol.^{32,33}

In agonism studies, responses to the cumulative addition of 5-HT or agonist were expressed as percentage relaxation of carbachol-induced tone. For each agonist, the negative logarithm of the molar concentration that relaxed 50% relative to their individual maximum effect (pEC₅₀) was calculated. The intrinsic activity (i.a.) versus 5-HT was determined comparing the maximum relaxation obtained at the concentration of 1 or 3 μM in the same TMM preparation.

In antagonism studies, the effect of the antagonists on 5-HT pEC₅₀ values was registered, and the values were compared by using the unpaired Student's *t*-test. A *p* value of 0.05 or less was considered statistically significant. The apparent affinity (pK_b) for antagonists against 5-HT was determined.

For each compound, the ability to displace the specific ligands [³H]LY 278584 and [³H]raclopride from 5-HT₃ sites of rat entorhinal cortex or dopamine D₂ sites of rat striatum, respectively, was also determined.^{34,35} No functional studies were carried out with the compounds that exhibited affinity at these receptors.

Table 1. 2,3-Dihydro-2-oxo-1H-benzimidazole-1-carboxamides

compd	R	synthetic method	mp, °C	formula	5-HT ₄ binding K_i (nM) \pm SE ^a	5-HT ₃ binding IC ₅₀ (nM)	5-HT ₄ antagonism $pK_b \pm$ SE ^a	5-HT ₄ agonism i.a.
1	H	A	188–190	C ₁₅ H ₂₁ N ₅ O ₂	\approx 1000 ^b	>1000	<6.0	0.35
2	Me	A	87–90	C ₁₆ H ₂₃ N ₅ O ₂	305.8 \pm 20.8	>1000	<6.0	0.42
3a	Et	A	194–196 ^c	C ₁₇ H ₂₅ N ₅ O ₂ ·2C ₄ H ₄ O ₄	75.4 \pm 19.9	>1000	6.79 \pm 0.12	<0.30
4	<i>n</i> -Pr	A	205–208 ^c	C ₁₈ H ₂₇ N ₅ O ₂	252.9 \pm 54.4	>1000	<6.0	0.48
5a	<i>i</i> -Pr	A	126–128	C ₁₈ H ₂₇ N ₅ O ₂	91.1 \pm 10.8	>1000	<6.0	0.94
6a	<i>c</i> -Pr	C	104–108	C ₁₈ H ₂₅ N ₅ O ₂	10.2 \pm 3.2	>1000	7.24 \pm 0.18	0.36
7	<i>n</i> -Bu	A	202–204 ^c	C ₁₉ H ₂₉ N ₅ O ₂ ·2C ₄ H ₄ O ₄	>1000 ^b	>1000	<6.0	<0.30
8	allyl	A	198–201 ^c	C ₁₈ H ₂₅ N ₅ O ₂ ·2C ₄ H ₄ O ₄	>1000 ^b	>1000	<6.0	0.63
9	PhCH ₂	C	110–113	C ₂₂ H ₂₇ N ₅ O ₂	>1000 ^b	>1000	<6.0	<0.30
10	Ph(CH ₂) ₂	A	212 dec ^d	C ₂₃ H ₂₉ N ₅ O ₂ ·2HCl	>1000 ^b	>1000	<6.0	<0.30
BIMU 8					63.9 \pm 14.8	0.5 \pm 0.1 ^e	<6.0	0.70
GR 113808					0.11 \pm 0.01	>1000	9.49 \pm 0.06	NT

^a Standard error. ^b IC₅₀ (nM). ^c Fumarate. ^d Hydrochloride. ^e K_i (nM).

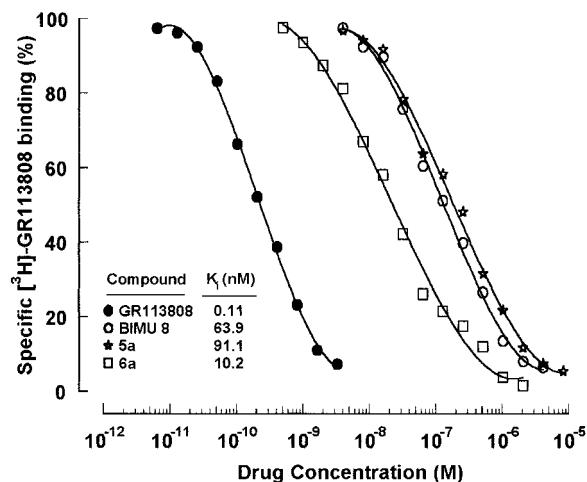


Figure 4. Competition for [³H]GR 113808 binding sites in guinea pig striatum by compounds **5a**, **6a**, GR 113808, and BIMU 8.

Results and Discussion

Pharmacological results are reported in Tables 1–4.

To study the effect of the 3-substituent of the benzimidazolone ring and because of the synthetic availability, we fixed the 4-methylpiperazine moiety and a two-methylene spacer as our starting point. Examination of the results in Table 1 showed that the nature of the substituent in the 3-position of the benzimidazolone ring was critical for 5-HT₄ receptor affinity. Only compounds with ethyl (**3a**, K_i = 75.4 nM), isopropyl (**5a**, K_i = 91.1 nM), and cyclopropyl (**6a**, K_i = 10.2 nM) substituents showed moderate to high affinity for the 5-HT₄ receptor, and **6a** was even more potent than the reference benzimidazolone BIMU 8 (K_i = 63.9 nM), although it showed weaker affinity than GR 113808 (K_i = 0.11 nM) (Figure 4). Replacement by other small groups such as hydrogen (**1**, IC₅₀ \approx 1 μ M), methyl (**2**, K_i = 305.8 nM), or propyl (**4**, K_i = 252.9 nM) led to a large decrease in affinity. In addition, the introduction of bulkier groups, e.g., butyl (**7**), benzyl (**9**), phenylethyl (**10**), severely reduced 5-HT₄ receptor affinity (IC₅₀ > 1 μ M).

All compounds showed a remarkable improvement in selectivity versus 5-HT₃ receptor (IC₅₀ > 1 μ M) with

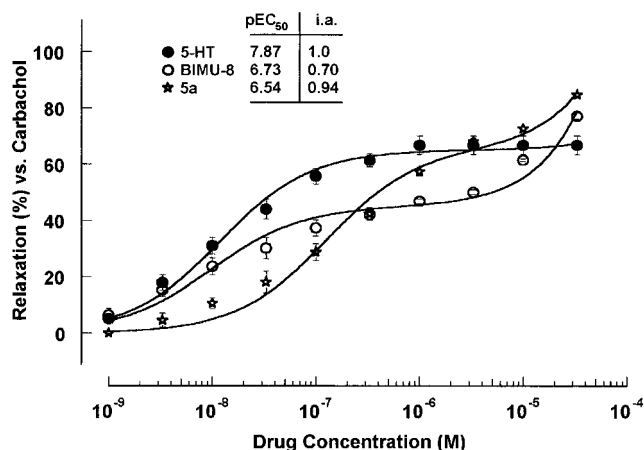
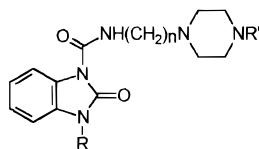


Figure 5. Concentration–relaxation curves to 5-HT, BIMU 8, and compound **5a** in rat isolated TMM. Values are mean, with SEM, n = 2–3.

respect to the reference compound BIMU 8. It can therefore be concluded that in the benzimidazolone series replacement of the rigid tropane ring (BIMU 8 and BIMU 1) by a flexible piperazine moiety significantly reduced or abolished 5-HT₃ affinity while retained the 5-HT₄ affinity. No compound displayed significant affinity for the D₂ receptor (IC₅₀ > 1 μ M, data not shown).

The 5-HT₄ receptor activity was also dependent on the substituent at the 3-position of the benzimidazolone ring. Ethyl and cyclopropyl derivatives **3a** and **6a** showed moderate antagonist activity (pK_b = 6.79 and 7.24, respectively). However, isopropyl derivative **5a** and BIMU 8 acted as partial agonists, showing a classical biphasic concentration–relaxation curve with pEC₅₀ values of 6.54 and 6.73 and with an intrinsic activity of 0.94 and 0.70, respectively (Figure 5). At concentrations higher than 1 μ M, both compounds showed 5-HT₄-independent relaxation effects, probably due to a muscarinic cholinergic blockade.³³ The 5-HT₄ antagonist GR 113808 induced a parallel, dextral shift of the concentration–relaxation curves for 5-HT, **5a**, and BIMU 8 in the rat TMM with pK_b values from 9.1 to 9.5 (data not shown). The observed change on the pharmacological profile of benzimidazolones **1–10** sug-

Table 2. 2,3-Dihydro-2-oxo-1*H*-benzimidazole-1-carboxamides: 3-Ethyl and 3-Cyclopropyl Derivatives

compd	R	n	R'	synthetic method	mp, °C	formula	5-HT ₄ binding K_i (nM) \pm SE ^a	5-HT ₃ binding IC ₅₀ (nM)	5-HT ₄ antagonism $pK_b \pm$ SE ^a	5-HT ₄ agonism i.a.
3a	Et	2	Me	A	194–196 ^b	C ₁₇ H ₂₇ N ₅ O ₂ ·2C ₄ H ₄ O ₄	75.4 \pm 19.9	>1000	6.79 \pm 0.12	<0.30
3b	Et	2	Et	C	90–93	C ₁₈ H ₂₇ N ₅ O ₂	17.9 \pm 1.9	>1000	6.89 \pm 0.14	<0.30
3c	Et	2	<i>n</i> -Pr	A	97–100	C ₁₉ H ₂₉ N ₅ O ₂	16.9 \pm 2.0	>1000	7.63 \pm 0.12	<0.30
3d	Et	2	<i>i</i> -Pr	C	208–210 ^b	C ₁₉ H ₂₉ N ₅ O ₂ ·2C ₄ H ₄ O ₄	25.2 \pm 2.5	>1000	7.42 \pm 0.10	<0.30
3e	Et	2	<i>n</i> -Bu	A	203–205 ^b	C ₂₀ H ₃₁ N ₅ O ₂ ·2C ₄ H ₄ O ₄	34.1 \pm 2.0	>1000	7.73 \pm 0.18	<0.30
6a	<i>c</i> -Pr	2	Me	C	104–108	C ₁₈ H ₂₅ N ₅ O ₂	10.2 \pm 3.2	>1000	7.24 \pm 0.18	0.36
6b	<i>c</i> -Pr	2	<i>n</i> -Pr	C	62–64	C ₂₀ H ₂₉ N ₅ O ₂	8.2 \pm 1.0	>1000	7.46 \pm 0.16	0.41
6c	<i>c</i> -Pr	2	<i>i</i> -Pr	B	oil	C ₂₀ H ₂₉ N ₅ O ₂	6.7 \pm 0.4	>1000	7.38 \pm 0.15	<0.30
6d	<i>c</i> -Pr	2	<i>n</i> -Bu	B	oil	C ₂₁ H ₃₁ N ₅ O ₂	13.3 \pm 0.9	>1000	7.51 \pm 0.20	0.46
6e	<i>c</i> -Pr	1	Me	C	86–88	C ₁₇ H ₂₃ N ₅ O ₂	49.4 \pm 8.2	>1000	6.19 \pm 0.10	0.53
6f	<i>c</i> -Pr	3	Me	B	>250 ^c	C ₁₉ H ₂₇ N ₅ O ₂ ·2HCl	39.7 \pm 7.4	>1000	6.40 \pm 0.09	0.46

^a Standard error. ^b Fumarate. ^c Hydrochloride.

gests that the mode of binding might not be identical through the series. A similar dramatic variation in the 5-HT₄ pharmacological activity due only to a small structural modification has been previously reported for benzoate derivatives; as a consequence, a hypothetical model for the 5-HT₄ receptor with two sites for the binding of agonist and antagonist molecules was proposed.¹⁷

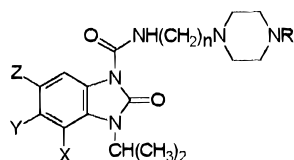
These findings led us to evaluate a series of 3-ethyl- and 3-cyclopropylbenzimidazolone derivatives (Table 2). According to previously proposed 5-HT₄ receptor models, a fairly large lipophilic region is located near the basic nitrogen recognition sites, and increasing the size and lipophilic character of the side chain may result in increased affinity.¹⁶ The first modification, therefore, was to increase the size of the 4-substituent of the piperazine moiety. Replacement of the methyl group in compound **3a** ($K_i = 75.4$ nM) by ethyl or propyl groups (**3b**, $K_i = 17.9$ nM; **3c**, $K_i = 16.9$ nM, respectively) led to an increased affinity for the 5-HT₄ receptor. However, the introduction of an isopropyl or a butyl group (**3d**, $K_i = 25.2$ nM; **3e**, $K_i = 34.1$ nM, respectively) did not improve the affinity of **3b**. In the cyclopropyl series the replacement of the methyl group in **6a** ($K_i = 10.2$ nM) by a propyl (**6b**, $K_i = 8.2$ nM), isopropyl (**6c**, $K_i = 6.7$ nM), or butyl (**6d**, $K_i = 13.3$ nM) group did not produce a significant change on the affinity for the 5-HT₄ receptor. These data also showed that a cyclopropyl substituent in the 3-position of the benzimidazolone ring is better than an ethyl group for 5-HT₄ receptor affinity. On the other side, these structural modifications had divergent and slight influence on 5-HT₄ antagonist activity. Although butyl derivatives **3e** and **6d** displayed moderate 5-HT₄ affinity, they showed the highest 5-HT₄ antagonist values ($pK_b = 7.73$, 7.51 respectively). This result agrees with the hypothesized positive steric interaction between the *n*-butyl group and the hydrophobic pocket of the 5-HT₄ receptor in compounds with 5-HT₄ antagonist activity.¹⁶ Besides, propyl derivatives **3c** and **6b** ($pK_b = 7.63$ and 7.46) did not show remarkable differences on the antagonist activity with respect to the related isopropyl derivatives **3d** and **6c** ($pK_b = 7.42$ and 7.38) and methyl analogues **3a** and **6a** ($pK_b = 6.79$ and 7.24). Moreover, improvement on the affinity did not always imply a correlative improvement on the

5-HT₄ antagonism, e.g., compare **3b** and **3e** or **6c** and **6d**. Since those compounds were approximately equipotent, antagonist activity for the 5-HT₄ receptor seems to be unrelated to the size of the 4-substituent of the piperazine moiety in the ethyl and cyclopropyl series.

The modification of the alkylene spacer was the next step to be studied. The shortening of the two-methylene chain of derivative **6a** led to a decrease in affinity (**6e**, $K_i = 49.4$ nM) and also reduced the antagonist activity ($pK_b = 6.19$). On the other hand, the lengthening to a three-methylene chain caused a similar reduction in both affinity and antagonist activity (**6f**, $K_i = 39.7$ nM, $pK_b = 6.40$). Consequently, in this series a chain length of two methylenes seem to be optimal for high affinity and antagonist activity at the 5-HT₄ receptor site.

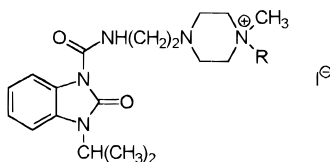
Similar structural modifications were also carried out on the 3-isopropyl derivative **5a** (Table 3). As a general trend a significant reduction in 5-HT₄ receptor affinity with regard to the corresponding 3-ethyl and 3-cyclopropyl derivatives was observed, with a partial agonist profile comparable to that of the reference compound BIMU 8 ($K_i = 63.9$ nM, i.a. = 0.70). Surprisingly, butyl derivative **5h** ($K_i = 38.9$ nM, $pK_b = 7.94$) showed the most potent 5-HT₄ antagonist activity in our series, which confirms the key role of the butyl group in the antagonist profile for the 5-HT₄ receptor.

Replacement in derivative **5a** of the methyl substituent at the 4-position of the piperazine by ethyl (**5e**, $K_i = 103.6$ nM, i.a. = 0.87), propyl (**5f**, $K_i = 87.3$ nM, i.a. = 0.68), or isopropyl (**5g**, $K_i = 89.8$ nM, i.a. = 0.66) substituents retained affinity for the 5-HT₄ receptor and caused a reduction in partial agonist activity. 4-Fluorophenoxypipyl, 4-fluorophenylmethyl, and piperonyl groups have shown to be excellent substituents for 5-HT₄ ligands with agonist activity (Figure 2). The introduction of these substituents at the 4-position of the piperazine led to **5i** ($K_i = 375$ nM, i.a. = 0.78), **5j** ($K_i = 262$ nM, i.a. = 0.78), and **5k** (IC₅₀ > 1 μ M, i.a. = 0.30) derivatives with reduced affinity for the 5-HT₄ receptor and a decrease in the partial agonist activity. These data suggest the existence of steric limitations around the 4-substituent of the piperazine moiety for 5-HT₄ agonism and that a methyl group seems to be optimal in terms of both affinity and agonist activity at the 5-HT₄ receptor site.

Table 3. 2,3-Dihydro-2-oxo-1H-benzimidazole-1-carboxamides: 3-Isopropyl Derivatives

compd	X	Y	Z	n	R'	synthetic method	mp, °C	formula	5-HT ₄ binding K _i (nM) ± SE ^a	5-HT ₃ binding IC ₅₀ (nM)	5-HT ₄ antagonism pK _b ± SE ^a	5-HT ₄ agonism i.a.
5a	H	H	H	2	Me	A	126–128	C ₁₈ H ₂₇ N ₅ O ₂	91.1 ± 10.8	>1000	<6.0	0.94
5b	H	H	H	0	Me	A	141–143	C ₁₆ H ₂₃ N ₅ O ₂	>1000 ^b	200.8 ^c	<6.0	<0.30
5c	H	H	H	1	Me	C	71–73	C ₁₇ H ₂₅ N ₅ O ₂	>1000 ^b	309.0 ^c	<6.0	<0.30
5d	H	H	H	3	Me	A	>240 ^d	C ₁₉ H ₂₉ N ₅ O ₂ ·2HCl	>1000 ^b	>1000	<6.0	<0.30
5e	H	H	H	2	Et	C	190–193 ^e	C ₁₉ H ₂₉ N ₅ O ₂ ·2C ₄ H ₄ O ₄	103.6 ± 7.8	>1000	6.62 ± 0.13	0.87
5f	H	H	H	2	<i>n</i> -Pr	C	202–205 ^e	C ₂₀ H ₃₁ N ₅ O ₂ ·2C ₄ H ₄ O ₄	87.3 ± 10.3	>1000	6.89 ± 0.19	0.68
5g	H	H	H	2	<i>i</i> -Pr	C	210–212 ^e	C ₂₀ H ₃₁ N ₅ O ₂ ·2C ₄ H ₄ O ₄	89.8 ± 6.0	>1000	<6.0	0.66
5h	H	H	H	2	<i>n</i> -Bu	A	210–212 ^e	C ₂₁ H ₃₃ N ₅ O ₂ ·2C ₄ H ₄ O ₄	38.9 ± 3.6	>1000	7.94 ± 0.16	<0.30
5i	H	H	H	2	4-FPhO(CH ₂) ₃	B	90–94	C ₂₆ H ₃₄ FN ₅ O ₃	375.0 ± 23.7	>1000	6.11 ± 0.15	0.78
5j	H	H	H	2	4-FPhCH ₂	A	206–208 ^e	C ₂₄ H ₃₀ FN ₅ O ₂ ·2C ₄ H ₄ O ₄	262.0 ± 17.9	>1000	<6.0	0.78
5k	H	H	H	2	piperonyl	C	218 dec ^e	C ₂₅ H ₃₁ N ₅ O ₄ ·2C ₄ H ₄ O ₄	>1000 ^b	>1000	<6.0	0.30
5l	H	H	Cl	2	Me	C	134–137	C ₁₈ H ₂₆ ClN ₅ O ₂	>1000 ^b	>1000	<6.0	0.70
5m	H	H	F	2	Me	C	130–134	C ₁₈ H ₂₆ FN ₅ O ₂	108.4 ± 11.0	>1000	6.88 ± 0.11	0.67
5n	H	H	MeO	2	Me	B	109–111	C ₁₉ H ₂₉ N ₅ O ₃	>1000 ^b	>1000	<6.0	<0.30
5o	H	NO ₂	F	2	Me	C	198–201	C ₁₈ H ₂₅ FN ₆ O ₄	>1000 ^b	>1000	<6.0	<0.30
5p	H	NH ₂	F	2	Me	f	84 dec	C ₁₈ H ₂₇ FN ₆ O ₂	>1000 ^b	>1000	<6.0	<0.30
5q	Me	H	H	2	Me	B	82–84	C ₁₉ H ₂₉ N ₅ O ₂	216.4 ± 15.3	73.9 ^c	<6.0	<0.30

^a Standard error. ^b IC₅₀ (nM). ^c K_i (nM). ^d Hydrochloride. ^e Fumarate. ^f Reduction of 5o with Zn/1 N NaOH.

Table 4. 2,3-Dihydro-2-oxo-1H-benzimidazole-1-carboxamides: Quaternary Derivatives

compd	R	synthetic method	mp, °C	formula	5-HT ₄ binding K _i (nM) ± SE	5-HT ₃ binding IC ₅₀ (nM)	5-HT ₄ antagonism pK _b ± SE	5-HT ₄ agonism i.a.
5r	<i>n</i> -Bu	D	170–172	C ₂₂ H ₃₆ IN ₅ O ₂	66.3 ± 4.6	>1000	6.0 ± 0.23	0.93
5s	Me	D	60 dec	C ₁₉ H ₃₀ IN ₅ O ₂	68.5 ± 26.8	>1000	<6.0	0.75
5t	allyl	E	177–179	C ₂₁ H ₃₂ IN ₅ O ₂	56.3 ± 6.00	>1000	6.28 ± 0.17	<0.30

Modifications of the spacer in the parent compound 5a led to derivatives 5b, 5c, and 5d with a remarkable decrease in the 5-HT₄ affinity (IC₅₀ > 1 μM) and slight reduction in the 5-HT₄ versus 5-HT₃ receptor selectivity. As observed previously for ethyl and cyclopropyl derivatives, the length of the alkylene chain is an important structural feature in recognition by the 5-HT₄ receptor binding site. A two-methylene spacer is also optimal for high 5-HT₄ agonist activity.

Introduction of substituents on the benzene ring had a negative effect on 5-HT₄ receptor affinity, except for derivative 5m (K_i = 108.4 nM) which has a fluorine in the 6-position. The structurally related benzimidazolone DAU 6285 has a methoxy group in the 6-position and behaves as a 5-HT₄ antagonist.²⁰ A similar substitution in this series led to derivative 5n (IC₅₀ > 1 μM) with a remarkable reduction in the affinity for the 5-HT₄ receptor. It can be concluded that the 5-HT₄ receptor can only accommodate small substituents in the position of the benzene ring in our series. Therefore, it seems that the electronic distribution within the benzimidazole ring has a small effect on the 5-HT₄ receptor affinity, whereas steric hindrance might play a significant role.

It has been reported that increasing the polarity by *n*-butyl quaternization of renzapride increases the 5-HT₄ receptor agonist potency.³⁶ Several quaternary deriva-

tives of 5a were prepared (Table 4). All of them showed higher affinity for the 5-HT₄ receptor (K_i = 56.3–68.5 nM) than 5a, but while the butyl derivative 5r (i.a. = 0.93) retained the partial agonist activity, the methyl derivative 5s (i.a. = 0.75) showed a significant decrease in the partial agonist activity and the allyl derivative 5t (i.a. < 0.30) was only weakly active as a 5-HT₄ agonist.

Conclusion

1. The 2,3-dihydro-2-oxo-1H-benzimidazole-1-carboxamides including a piperazine moiety described in this paper represent a novel class of ligands at the 5-HT₄ receptor with selectivity over 5-HT₃ and D₂ receptors.

2. The 3-substituent at the benzimidazolone ring is a determinant structural feature for both binding and pharmacological profile. 3-Ethyl and 3-cyclopropyl derivatives showed high affinity (6c, K_i = 6.7 nM) and moderate antagonist activity (3e, pK_b = 7.73), whereas 3-isopropyl analogues exhibited moderate affinity and high partial agonist activity (5a, K_i = 91.1 nM, i.a. = 0.94). The reversal of the pharmacological activity due only to a small structural difference might confirm the existence of two binding sites at the 5-HT₄ receptor.

3. The length of the alkylene spacer is an important structural feature in the recognition by the 5-HT₄

receptor binding site. A chain length of two methylenes is optimal for high affinity and also for good antagonist or agonist activity.

4. In the ethyl and cyclopropyl series, 5-HT₄ antagonist activity seems to be unrelated to the nature of the 4-substituent of the piperazine moiety. However, the presence of a butyl substituent is a favorable pattern for 5-HT₄ antagonism and even in the 5-HT₄ agonist series (isopropyl derivatives) caused a reversal of the pharmacological profile (**5h**, p*K*_b = 7.94).

5. Derivative **5a** showed a partial agonist activity greater than that of BIMU 8 and exhibited better selectivity versus 5-HT₃ and D₂ receptors.

6. *N*-Butyl quaternization of **5a** led to **5r** (*K*_i = 66.3, i.a. = 0.93) with an improvement in affinity for the 5-HT₄ receptor and a similar partial agonist activity.

Experimental Section

Chemistry. Flash column chromatography was performed on silica gel, particle size 60 Å, mesh = 230–400 (Merck). Melting points were determined in open capillary tubes on a Büchi SMP-20 apparatus and are uncorrected. Elemental analyses are within ±0.4% of the theoretical values. IR spectra were taken on a Perkin-Elmer 1310 instrument on KBr plates. ¹H and ¹³C NMR were recorded on a Bruker AC-200 spectrometer; chemical shifts δ are reported in parts per million (ppm) downfield from tetramethylsilane (TMS), which was used as an internal standard. Spectral data are consistent with assigned structures.

***N*-(1-Methylethyl)-2-nitroaniline, 11a.** 2-Nitroaniline (40.0 g, 0.29 mol), 2,2-dimethoxypropane (54.5 g, 0.52 mol), and TFA (41.5 g, 0.29 mol) were dissolved in toluene (600 mL) and stirred at room temperature for 1 h. A boron–pyridine complex (26.6 g, 0.29 mol) was slowly added. The reaction mixture was stirred for 20 h. The solvent was evaporated in vacuo, and the residue was taken up into water and extracted with CH₂Cl₂. The organic extract was dried (Na₂SO₄) and concentrated in vacuo. The residue was purified by flash chromatography on silica gel using hexane and ethyl acetate (95/5) to give **11a** as an orange oil (44.0 g, 84%). ¹H NMR (CDCl₃): δ 8.19 (1H, d); 7.41 (1H, t); 6.83 (1H, d); 6.49 (1H, t); 3.81 (1H, m); 1.28 (6H, d). ¹³C NMR (CDCl₃): δ 144.7, 136.0, 131.6, 127.0, 114.8, 114.0, 43.8, 22.6.

***N*-Cyclopropyl-2-nitroaniline, 11b.** A mixture of 2-chloronitrobenzene (3.1 g, 20 mmol) and cyclopropylamine (3.5 mL, 50 mmol) was placed in a high-pressure vessel and heated at 100 °C for 24 h. Then the reactor was opened, the reaction mixture was diluted with water and extracted with CH₂Cl₂, and the extract was washed with water and dried over Na₂SO₄. The solvent was evaporated in vacuo, and the residue was purified by flash chromatography using hexane and CH₂Cl₂ (7/3) as eluent to give **11b** as an orange oil (2.1 g, 60%). ¹H NMR (CDCl₃): δ 8.20 (1H, d); 7.61–7.42 (2H, m); 7.30 (1H, dd); 6.72 (1H, m); 2.60 (1H, m); 0.81–1.01 (2H, m); 0.60–0.72 (2H, m). ¹³C NMR (CDCl₃): δ 164.0, 146.0, 135.8, 126.4, 115.8, 115.2, 24.4, 7.7.

2-[(1-Methylethyl)amino]aniline, 12a. A solution of **11a** (2.3 g, 12.7 mmol) in EtOH (50 mL) was hydrogenated over 10% palladium/carbon (0.5 g) at 3 atm and 20 °C for 16 h. The catalyst was filtered off and the filtrate evaporated in vacuo to give **12a** (1.6 g, 83%). ¹H NMR (CDCl₃): δ 6.71–6.62 (4H, m); 3.61 (1H, m); 3.13 (3H, br s); 1.20 (6H, d). ¹³C NMR (CDCl₃): δ 136.7, 134.4, 120.6, 118.4, 116.7, 112.9, 44.3, 23.1.

5-Chloro-2-[(1-methylethyl)amino]aniline, 12c. A suspension of **11c** (20.0 g, 93 mmol) in EtOH (400 mL) and 1 N NaOH (300 mL) was stirred vigorously and heated until the solution boiled gently. The bath was removed, and zinc dust (60.8 g, 930 mmol) was added in several portions frequently enough to keep the solution boiling. After refluxing for 1 h, the mixture was filtered and the filtrate was poured into water (100 mL) and extracted with CH₂Cl₂. The organic phase was dried with MgSO₄ and evaporated in vacuo to give **12c** (17.0

g, 99%). ¹H NMR (CDCl₃): δ 6.78 (1H, br d); 6.65 (1H, br s); 6.57 (1H, d); 3.52 (1H, m); 3.38 (3H, br s); 1.22 (6H, d). ¹³C NMR (CDCl₃): δ 135.9, 135.2, 122.9, 119.8, 116.2, 113.7, 44.3, 22.9.

1,3-Dihydro-1-(1-methylethyl)-2H-benzimidazol-2-one, 13a. A solution of **12a** (29.6 g, 0.19 mol) and *N,N*-carbonyldiimidazole (32.1 g, 0.19 mol) in dry THF (400 mL) was stirred at room temperature for 20 h and then evaporated. The residue was taken up in water and extracted with CH₂Cl₂. The dried organic phase was evaporated, and the residue was purified by flash chromatography using CH₂Cl₂ and methanol (98/2) as eluent to give a product that was triturated with hexane and filtered to yield **13a** as pale-yellow solid (16.1 g, 46%). Mp: 125–127 °C. ¹H NMR (CDCl₃): δ 7.22–7.15 (4H, m); 4.78 (1H, m); 1.58 (6H, d). ¹³C NMR (CDCl₃): δ 155.4, 128.9, 128.3, 121.0, 120.8, 109.9, 109.2, 44.6, 20.2.

1,3-Dihydro-5-fluoro-1-(1-methylethyl)-6-nitro-2H-benzimidazol-2-one, 13f. The compound **13d** (11.0 g, 56.7 mmol) was suspended in acetic anhydride (75 mL) and cooled in an ice bath. To this suspension was added dropwise 22.5 mL of a mixture of nitric acid and acetic anhydride (1/2). The reaction mixture was stirred for 3 min before being poured into ice-water. The solid was filtered, washed with water and cold Et₂O, and dried to afford **13f** (11.8 g, 87%). ¹H NMR (CDCl₃): δ 11.20 (1H, br s); 7.62 (1H, d); 6.80 (1H, d); 4.58 (1H, m); 1.41 (6H, d). ¹³C NMR (CDCl₃): δ 154.7, 152.7 (d, *J* = 245 Hz), 134.7 (d, *J* = 12 Hz), 130.2 (d, *J* = 7 Hz), 124.9, 104.9, 98.5 (d, *J* = 27 Hz), 45.0, 19.9.

1,3-Dihydro-1-propyl-2H-benzimidazol-2-one, 13h. HNa (2.0 g, 50 mmol, 60% in mineral oil) was added portionwise to a stirred solution of 1,3-dihydro-2H-benzimidazol-2-one (6.7 g, 50 mmol) in dry DMF (50 mL). After 1 h, a solution of propyl iodide (4.9 g, 50 mmol) in dry DMF (20 mL) was added dropwise and the mixture stirred at room temperature for 24 h. The solvent was removed in vacuo, and the residue was taken up into water and extracted with CH₂Cl₂. The organic phase was dried over Na₂SO₄, the solvent was evaporated to dryness, and the residue was chromatographed on a column of silica gel using a mixture of dichloromethane and methanol (98/2) as eluent to separate **13h** (55%). Mp: 97–9 °C. ¹H NMR (CDCl₃): δ 10.38 (1H, br s); 7.20–6.94 (4H, m); 3.90 (2H, t); 1.83 (2H, m); 1.02 (3H, t). ¹³C NMR (CDCl₃): δ 155.8, 130.4, 128.0, 121.3, 121.1, 109.6, 107.8, 42.4, 21.7, 11.3.

2,3-Dihydro-2-oxo-1H-benzimidazole-1-carboxylic Acid, 1,1-Dimethylethyl Ester, 14. HNa (1.6 g, 41 mmol, 60% in mineral oil) was added portionwise to a stirred solution of 1,3-dihydro-2H-benzimidazol-2-one (5.0 g, 37.2 mmol) in dry DMF (100 mL) under an atmosphere of argon. After 1.5 h, a solution of di-*tert*-butyl dicarbonate (8.1 g, 37.2 mmol) in dry DMF (20 mL) was added dropwise and the mixture stirred at room temperature for 24 h. The solvent was evaporated and the residue diluted with saturated NH₄Cl solution and extracted with EtOAc. The residue was purified by flash chromatography using hexane and EtOAc (7/3) as eluent to give **14** as a white solid (6.6 g, 76%). Mp: >250 °C. ¹H NMR (CDCl₃): δ 10.44 (1H, s); 7.69 (1H, d); 7.18–7.06 (3H, m); 1.68 (9H, s). ¹³C NMR (CDCl₃): δ 153.4, 148.6, 127.4, 126.8, 124.1, 122.0, 114.4, 110.0, 85.0, 28.1.

1,3-Dihydro-1-(2-phenylethyl)-2H-benzimidazol-2-one, 13l. A mixture of **14** (2.3 g, 10 mmol), 2-phenylethyl bromide (1.8 g, 10 mmol), K₂CO₃ (4.1 g, 30 mmol), and DMF (40 mL) was stirred at reflux for 24 h. The mixture was concentrated in vacuo and the residue extracted with CH₂Cl₂. The organic phase was dried over Na₂SO₄, and the solvent was evaporated in vacuo. The residue was purified by flash chromatography using a mixture of hexane and ethyl acetate (7/3) as eluent to yield **13l** (90%). Mp: 142–144 °C. ¹H NMR (DMSO-*d*₆): δ 10.40 (1H, br s); 7.18–7.01 (4H, m); 7.01–6.81 (5H, m); 4.18 (2H, t); 3.14 (2H, t). ¹³C NMR (DMSO-*d*₆): δ 155.6, 138.2, 130.1, 128.8, 128.6, 128.0, 126.6, 121.4, 121.1, 109.7, 107.7, 42.4, 34.7.

Ethyl 4-(1-Methylethyl)piperazine-1-carboxylate, 15a. A mixture of 1-ethoxycarbonylpiperazine (7.9 g, 50 mmol), isopropyl iodide (5 mL, 50 mmol), K₂CO₃ (8.3 g, 60 mmol), and

CH₃CN (100 mL) was refluxed for 24 h. The solvent was evaporated in vacuo, and the residue was poured onto water and extracted with dichloromethane. The organic phase was dried over Na₂SO₄ and the solvent evaporated in vacuo to yield **15a** as an oil (98%). ¹H NMR (CDCl₃): δ 4.12 (2H, q); 3.48 (4H, m); 2.70 (1H, m); 2.46 (4H, m); 1.25 (3H, t); 1.04 (6H, d). ¹³C NMR (CDCl₃): δ 154.8, 61.1, 54.5, 48.3, 44.0, 18.3, 14.6.

1-(1-Methylethyl)piperazine 16a. A solution of **15a** (9.8 g, 49 mmol) and 45% HBr (200 mL) was heated at reflux for 3 h. The reaction mixture was cooled, 50% NaOH solution was added until pH > 7, and the mixture was extracted with a solution of CH₃Cl and *i*-PrOH (4/1). The organic phase was dried over Na₂SO₄, and the solvent was evaporated in vacuo to yield **16a** (90%) as an oil. ¹H NMR (CDCl₃): δ 2.90 (4H, m); 2.63 (1H, m); 2.48 (4H, m); 1.21 (1H, br s); 1.04 (6H, d). ¹³C NMR (CDCl₃): δ 54.8, 49.9, 46.4, 18.3.

1-Butylpiperazine (Hydrobromide), 16b. A mixture of **15b** (1.0 g, 4.6 mmol) and 45% HBr (30 mL) was heated at reflux for 1 h. The solvent was evaporated and the residue triturated with Et₂O and filtered to give **16b** (1.4 g, 98%). Mp: >240 °C. ¹H NMR (DMSO-*d*₆ + D₂O): δ 4.01 (5H, m); 3.41 (6H, m); 3.12 (2H, m); 1.60 (2H, m); 1.28 (2H, m); 0.86 (3H, t). ¹³C NMR (DMSO-*d*₆ + D₂O): δ 56.6, 48.4, 40.7, 25.6, 19.8, 14.0.

1-(4-Fluorophenylmethyl)piperazine, 16c. The compound was prepared according to the process reported for **16b**. The hydrobromide was dissolved in a solution of 10% NaOH, extracted with CH₂Cl₂, and dried over Na₂SO₄. Evaporation of solvent gave **16c** as an oil. Yield: 99%. ¹H NMR (CDCl₃): δ 7.30–6.98 (4H, m); 3.45 (2H, s); 2.86 (4H, m); 2.38 (4H, m); 1.80 (1H, s). ¹³C NMR (CDCl₃): δ 161.7 (d, *J* = 243 Hz), 133.6 (d, *J* = 3 Hz), 130.4 (d, *J* = 7 Hz), 114.7 (d, *J* = 21 Hz), 62.6, 54.2, 45.9.

(4-Methylpiperazin-1-yl)acetonitrile, 17a. A suspension of 1-methylpiperazine (9.0 g, 90 mmol), acetonitrile (60 mL), K₂CO₃ (60.0 g, 430.0 mmol), and 2-chloroacetonitrile (8.2 g, 100.0 mmol) was stirred at room temperature for 20 h. Et₂O (100 mL) was added, and the suspension was filtered. The filtrate was concentrated in vacuo to give **17a** as a yellow solid (12.0 g, 96%). Mp: 50–52 °C. ¹H NMR (CDCl₃): δ 3.40 (2H, s); 2.63 (4H, m); 2.44 (4H, m); 2.25 (3H, s). ¹³C NMR (CDCl₃): δ 114.5, 54.3, 51.5, 45.6, 45.5.

(4-Propylpiperazin-1-yl)acetonitrile, 17b. A mixture of propylpiperazine dihydrobromide (10.0 g, 34.4 mmol), acetonitrile (100 mL), K₂CO₃ (15.2 g, 110.2 mmol), and chloroacetonitrile (2.8 g, 37.7 mmol) was stirred at reflux for 5 h. The solvent was evaporated, and the residue was taken up into water and extracted with Cl₂CH₂. The organic extract was dried over Na₂SO₄ and concentrated in vacuo to give **17b** as an oil (4.8 g, 83%). ¹H NMR (CDCl₃): δ 3.55 (2H, s); 2.66–2.26 (10H, m); 1.51 (2H, m); 0.90 (3H, t). ¹³C NMR (CDCl₃): δ 114.5, 60.0, 52.3, 51.5, 45.6, 19.7, 11.6.

2-(4-Methylpiperazin-1-yl)ethylamine, 18a. To a stirred suspension of H₄LiAl (3.3 g) in dry Et₂O (100 mL), cooled at 0 °C, was slowly added a solution of **17a** (11.2 g, 80.4 mmol) in dry Et₂O (150 mL) and dry THF (150 mL). The mixture was stirred at room temperature for 24 h, cooled in an ice bath, and hydrolyzed with a 20% NaOH solution (50 mL). After stirring for 20 min, the mixture was filtered and the solvent evaporated. The residue was dissolved in Et₂O and dried over Na₂SO₄. The solvent was evaporated to dryness to give **18a** as an oil (11.5 g, 99%). ¹H NMR (CDCl₃): δ 2.80 (2H, t); 2.45 (10H, m); 2.25 (3H, s); 1.61 (2H, br s). ¹³C NMR (CDCl₃): δ 60.9, 54.9, 52.9, 45.8, 38.5.

2,3-Dihydro-*N*-[2-(4-methylpiperazin-1-yl)ethyl]-2-oxo-1H-benzimidazole-1-carboxamide, 1 (Method A). Trichloromethyl chloroformate (5.5 g, 27.8 mmol) dissolved in dry THF (10 mL) was added dropwise to a stirred suspension of 1,3-dihydro-1H-benzimidazol-2-one (5.0 g, 37.2 mmol), activated charcoal (0.08 g), and dry THF (100 mL). The reaction mixture was refluxed for 8 h; then heating was removed while stirring was continued 15 h more. The solid was removed by filtration. Evaporation of the solvent led to a crude material which was triturated with Et₂O to give 2,3-dihydro-2-oxo-1H-

benzimidazole-1-carbonyl chloride (**19a**) (5.5 g). The solid was added to a solution of 2-(4-methylpiperazin-1-yl)ethylamine (**18a**) (2.8 g, 20.0 mmol) and triethylamine (2.8 g, 28.0 mmol) in dry THF (60 mL) and was stirred at room temperature for 24 h. The solvent was evaporated, and the residue was taken up into water and extracted with Cl₂CH₂. The organic extract was dried over Na₂SO₄ and concentrated in vacuo. The residue was purified by flash chromatography using CH₂Cl₂ and MeOH (9/1 to 3/1) as eluent to give **1** (2.4 g, 41%). Mp: 188–190 °C. ¹H NMR (CDCl₃): δ 8.18 (1H, m); 7.15 (3H, m); 3.48 (2H, t); 2.65–2.41 (10H, m); 2.25 (3H, s). ¹³C NMR (CDCl₃): δ 154.1, 151.9, 127.4, 123.5, 122.2, 114.9, 109.1, 56.4, 54.5, 52.2, 45.3, 36.5.

3-Cyclopropyl-2,3-dihydro-*N*-[2-[4-(1-methylethyl)piperazin-1-yl]ethyl]-2-oxo-1H-benzimidazole-1-carboxamide, 6c (Method B). 1-Cyclopropyl-1,3-dihydro-2H-benzimidazol-2-one (**13b**) (2.0 g, 11.5 mmol) was added to a suspension of HNa (60% mineral oil, 0.9 g, 23.0 mmol) in dry THF (30 mL). The mixture was stirred 15 min before adding a solution of phosgene in toluene (1.93 M) (23.8 mL, 46.0 mmol), and the stirring was continued for 1 h. The resulting mixture was filtered through Celite and concentrated (**19f**). The residue was dissolved in dry THF (50 mL), and 2-[4-(1-methylethyl)piperazin-1-yl]ethylamine (**18c**) (2.0 g, 11.6 mmol) and triethylamine (4 mL) were added. The solution was stirred at room temperature for 18 h. The solvent was evaporated, and the residue was taken up into water and extracted with Cl₂-CH₂. The organic extract was dried over Na₂SO₄ and concentrated in vacuo. The residue was purified by flash chromatography using CH₂Cl₂ and MeOH (95/5 to 9/1) as eluent to give **6c** as an oil (1.8 g, 42%). ¹H NMR (CDCl₃): δ 8.92 (1H, br s); 8.21 (1H, m); 7.20 (3H, m); 3.59 (2H, q); 2.90 (1H, m); 2.60 (11H, m); 1.11 (10H, m). ¹³C NMR (CDCl₃): δ 153.6, 151.7, 129.6, 126.3, 123.4, 122.7, 115.1, 108.5, 56.9, 54.4, 53.3, 48.6, 37.0, 22.4, 18.6, 6.0.

3-Cyclopropyl-2,3-dihydro-*N*-[2-(4-methylpiperazin-1-yl)ethyl]-2-oxo-1H-benzimidazole-1-carboxamide, 6a (Method C). 1-Cyclopropyl-1,3-dihydro-2H-benzimidazol-2-one (**13b**) (4.0 g, 23.0 mmol) was reflux with 2-chloroethyl isocyanate (2.9 g, 27.5 mmol) in toluene (50 mL) for 20 h. The mixture was cooled and the solid filtered and washed with hexane to give *N*-(2-chloroethyl)-3-cyclopropyl-2,3-dihydro-2-oxo-1H-benzimidazole-1-carboxamide (**20a**) (4.7 g, 73%). Mp: 152–153 °C. ¹H NMR (CDCl₃): δ 9.20 (1H, br s); 8.19 (1H, m); 7.30–7.17 (3H, m); 3.74 (4H, m); 2.93 (1H, m); 1.28–0.95 (4H, m). ¹³C NMR (CDCl₃): δ 153.6, 151.8, 129.6, 126.1, 123.7, 122.9, 115.1, 108.7, 43.2, 41.6, 22.5, 6.0.

A mixture of 1-methylpiperazine (0.8 g, 8.6 mmol), toluene (50 mL), **20a** (2.0 g, 7.1 mmol), K₂CO₃ (1.2 g, 8.6 mmol), and a pinch of KI was stirred at reflux for 17 h. The solvent was evaporated, and the residue was taken up into water and extracted with Cl₂CH₂. The organic extract was dried over Na₂SO₄ and concentrated in vacuo. The residue was purified by flash chromatography using CH₂Cl₂ and MeOH (9/1 to 3/1) as eluent to give **6a** (1.0 g, 41%). Mp: 104–108 °C. ¹H NMR (CDCl₃): δ 8.91 (1H, br s); 8.19 (1H, d); 7.20 (3H, m); 3.43 (2H, q); 2.80 (1H, m), 2.58 (10H, m); 2.22 (3H, s); 1.24–0.82 (4H, m). ¹³C NMR (CDCl₃): δ 153.2, 151.4, 129.3, 125.9, 123.1, 122.3, 114.7, 108.3, 56.5, 54.6, 52.5, 45.5, 36.7, 22.2, 5.8.

3-Cyclopropyl-2,3-dihydro-*N*-[4-methylpiperazin-1-yl]methyl]-2-oxo-1H-benzimidazole-1-carboxamide, 6e. 1-Cyclopropyl-1,3-dihydro-2H-benzimidazol-2-one (**13b**) (3.0 g, 17.2 mmol) was heated at 60 °C with chloromethyl isocyanate (1.9 g, 1.8 mL) in toluene (30 mL) for 24 h. The solvent was evaporated, and the residue was triturated with Et₂O to give *N*-(chloromethyl)-3-cyclopropyl-2,3-dihydro-2-oxo-1H-benzimidazole-1-carboxamide (**20d**) (4.0 g, 87%). Mp: 202–204 °C. ¹H NMR (DMSO-*d*₆): δ 9.23 (1H, br s); 8.16 (1H, d); 7.38–7.20 (3H, m); 4.79 (2H, d); 2.94 (1H, m); 1.21–0.82 (4H, m). ¹³C NMR (DMSO-*d*₆): δ 152.8, 151.0, 129.9, 125.8, 123.6, 122.2, 114.1, 109.1, 63.4, 22.4, 5.7.

A mixture of 1-methylpiperazine (0.56 g, 5.6 mmol), acetonitrile (25 mL), **20d** (1.0 g, 3.7 mmol), triethylamine (0.37 g, 3.7 mmol), and a pinch of KI was stirred at room temperature

for 10 min. The solvent was evaporated, and the residue was taken up into water and extracted with Cl_2CH_2 . The organic extract was dried over Na_2SO_4 and concentrated in vacuo. The residue was triturated with Et_2O to give **6e** (1.0 g, 81%). Mp: 86–88 °C. ^1H NMR (CDCl_3): δ 9.20 (1H, br s); 8.19 (1H, m); 7.21 (3H, m); 4.37 (2H, d); 2.91 (1H, m), 2.71 (4H, m); 2.49 (4H, m); 2.28 (3H, s); 1.10 (4H, m). ^{13}C NMR (CDCl_3): δ 153.4, 152.0, 129.4, 126.0, 123.5, 122.6, 115.0, 108.5, 61.3, 54.6, 49.1, 45.6, 22.3, 5.8.

5-Amino-2,3-dihydro-6-fluoro-3-(1-methylethyl)-*N*-[2-(4-methylpiperazin-1-yl)ethyl]-2-oxo-1*H*-benzimidazole-1-carboxamide, 5p. A suspension of **5o** (1.2 g, 3 mmol) in THF (75 mL) and 1 N NaOH (75 mL) was stirred vigorously and heated until the solution boiled gently. The bath was removed, and zinc dust (3.0 g, 45 mmol) was added in several portions frequently enough to keep the solution boiling. The mixture was refluxed for 1 h and filtered, and the filtrate was poured into water (50 mL) and extracted with CH_2Cl_2 . The organic phase was dried over MgSO_4 and evaporated in vacuo to give **5p** (0.95 g, 82%). Mp: 84 °C dec. ^1H NMR (CDCl_3): δ 6.78 (1H, d); 6.61 (1H, d); 5.63 (1H, br t); 4.60 (1H, m); 3.95 (2H, t); 3.61 (2H, br s); 3.45 (2H, q); 3.31 (4H, t); 2.30 (4H, t); 2.23 (3H, s); 1.41 (6H, d). ^{13}C NMR (CDCl_3): δ 157.6, 154.3, 147.5 (d, $J = 230$ Hz), 128.9 (d, $J = 14$ Hz), 124.0, 121.3 (d, $J = 11$ Hz), 98.4 (d, $J = 3$ Hz), 96.5 (d, $J = 26$ Hz), 54.4, 45.8, 44.8, 43.2, 40.5, 40.3, 19.9.

1-Butyl-4-[2-[2,3-dihydro-3-(1-methylethyl)-2-oxobenzimidazol-1-yloxoamino]ethyl]-1-methylpiperazonium Iodide, 5r. (Method D). A solution of *N*-[2-(4-butylpiperazin-1-yl)ethyl]-2,3-dihydro-3-(1-methylethyl)-2-oxo-1*H*-benzimidazole-1-carboxamide (**5h**) (1.2 g, 3.1 mmol) and methyl iodide (2 mL, 32.0 mmol) in THF (30 mL) was stirred at room temperature for 20 h. The solvent was removed, and the residue was taken up into water and extracted with Cl_2CH_2 . The organic extract was dried over Na_2SO_4 and concentrated in vacuo to give **5r** as a solid (1.1 g, 67%). Mp: 170–172 °C. ^1H NMR (CDCl_3): δ 9.12 (1H, br s); 8.21 (1H, m); 7.19 (3H, m); 4.70 (1H, m); 3.79 (4H, m); 3.61 (4H, m); 3.42 (3H, s); 3.05 (4H, m); 2.84 (2H, t); 1.84 (2H, m); 1.58 (8H, m); 1.03 (3H, t). ^{13}C NMR (CDCl_3): δ 152.5, 151.8, 127.5, 126.6, 123.3, 122.2, 115.1, 108.9, 60.6, 55.5, 47.5, 46.1, 45.3, 36.4, 23.9, 19.8, 19.5, 13.7.

1-Allyl-4-[2-[2,3-dihydro-3-(1-methylethyl)-2-oxobenzimidazol-1-yloxoamino]ethyl]-1-methylpiperazonium Iodide, 5t (Method E). To a cooled solution of 2,3-dihydro-3-(1-methylethyl)-*N*-[2-(4-methylpiperazin-1-yl)ethyl]-2-oxo-1*H*-benzimidazole-1-carboxamide (**5a**) (1.0 g, 2.9 mmol) in DMF (12 mL) was added allyl iodide (2.4 g, 14.3 mmol). The mixture was stirred at room temperature for 2 h. The solvent was evaporated and the residue purified by flash chromatography using CH_2Cl_2 and MeOH (9/1) as eluent to give **5t** as a white solid (1.0 g, 67%). Mp: 177–179 °C. ^1H NMR (CDCl_3): δ 9.08 (1H, br s); 8.25 (1H, m); 7.18 (3H, m); 4.18–5.77 (3H, m); 4.70 (1H, m); 4.51 (2H, d); 3.63 (6H, m); 3.44 (3H, s); 3.04 (4H, m); 2.83 (2H, t); 1.57 (6H, d). ^{13}C NMR (CDCl_3): δ 152.4, 151.8, 131.7, 127.2, 126.6, 123.3, 122.5, 115.4, 109.2, 65.8, 60.0, 54.8, 46.1, 45.0, 44.7, 35.1, 19.8.

Pharmacological Methods. 1. 5-HT₄ Receptor Binding Assay.³¹ Adult male Dunkin–Hartley guinea pigs weighing 350–400 g were used. Animals were killed by decapitation. The whole brain with the exception of the brainstem and cerebellum was quickly removed, and the various areas were dissected, weighed, and immediately frozen at –70 °C. Striatum used for the binding experiments was homogenized with an Ultra-Turrax homogenizer (setting 5 for 20 s) in 15 volumes of ice-cold 50 mM Hepes buffer (pH 7.4) and centrifuged at 48000*g* for 20 min (4 °C). The resulting pellet was resuspended in 10 volumes of ice-cold 50 mM Hepes buffer (pH 7.4) and was stored at –70 °C until use. At time of experiment, the membranes were diluted in the same ice-cold buffer (final dilution 1:80, wt/vol). Competition assays were performed in triplicate, in a final volume of 1 mL. To each assay tube were added the following: 0.1 mL of the displacer drug concentration (0.1 mL of vehicle if no competing drug was added) and

0.1 mL of [³H]GR 113808 (Amersham, 60–85 Ci/mmol) in buffer (final concentration 0.1 nM). Nonspecific binding was determined using 10 μM cold 5-HT. Binding experiment was initiated by addition of 0.8 mL of membrane suspension (800–900 μg of protein). After incubation for 30 min at 25 °C the reaction was stopped for vacuum filtration through Whatman GF/B glass pretreated filters (1% poly(ethylenimine) in 50 mM Hepes buffer), using a Brandel cell harvester. Filters were placed in scintillation poly(ethylene) vials (with 5 mL of scintillation cocktail) and equilibrated, and filter-retained radioactivity was measured in a liquid scintillation counter (Kontron Beta V). IC_{50} and K_i values were calculated using the computer program EBDA (McPherson).³⁷

2. 5-HT₃ Receptor Binding Assay.³⁴ Adult male Wistar rats weighing 220–280 g were used. Animals were killed by decapitation. The whole brain with the exception of the brainstem and cerebellum was quickly removed, and the various areas were dissected, weighed, and immediately frozen at –70 °C. Entorhinal cortex used for the binding experiments was homogenized with an Ultra-Turrax homogenizer (setting 5 for 20 s) in 10 volumes of ice-cold 0.32 M sucrose buffer and centrifuged at 1000*g* for 10 min (4 °C), and the supernatant was recentrifuged at 17000*g* for 20 min (4 °C). The resulting pellet was resuspended in 50 volumes of ice-cold 50 mM Tris-HCl buffer (pH 7.4), incubated at 37 °C for 10 min, and then centrifuged three times more at 48000*g* for 10 min (4 °C). The final pellet was resuspended in 10 volumes of ice-cold 50 mM Tris-HCl buffer containing 5 mM CaCl_2 and 0.1% ascorbate and was stored at –70 °C until use. At time of experiment, the membranes were diluted in the same ice-cold buffer with 10 μM pargyline (final dilution 1:40, wt/vol). Competition assays were performed in triplicate, in a final volume of 1 mL. To each assay tube were added the following: 0.1 mL of the displacer drug concentration (0.1 mL of vehicle if no competing drug was added) and 0.1 mL of [³H]LY 278584 (Amersham, 60–85 Ci/mmol) in buffer (final concentration 2 nM). Nonspecific binding was determined using 10 μM cold 5-HT. Binding experiment was initiated by addition of 0.8 mL of membrane suspension (500–600 μg of protein). After incubation for 30 min at 25 °C the reaction was stopped for vacuum filtration through Whatman GF/B glass pretreated filters (1% poly(ethylenimine) in 50 mM Tris-HCl buffer), using a Brandel cell harvester, followed by two washes with 5 mL of ice-cold 50 mM Tris-HCl (pH 7.4) buffer. Filters were placed in scintillation poly(ethylene) vials (with 5 mL of scintillation cocktail) and equilibrated, and filter-retained radioactivity was measured in a liquid scintillation counter (Kontron Beta V). IC_{50} and K_i values were calculated using the computer program EBDA (McPherson).³⁷

3. D₂ Receptor Binding Assay.³⁵ Adult male Wistar rats weighing 220–280 g were used. Animals were killed by decapitation. The whole brain with the exception of the brainstem and cerebellum was quickly removed, and the various areas were dissected, weighed, and immediately frozen at –70 °C. Striatum used for the binding experiments was homogenized with an Ultra-Turrax homogenizer (setting 5 for 20 s) in 50 volumes of ice-cold 50 mM Tris-HCl buffer (pH 7.7), centrifuged at 48000*g* for 10 min (4 °C). The resulting pellet was resuspended in 50 volumes of ice-cold 50 mM Tris-HCl buffer (pH 7.7), incubated at 37 °C for 10 min, and then centrifuged once more at 48000*g* for 10 min (4 °C). The resulting pellet was resuspended in 10 volumes of ice-cold 50 mM Tris-HCl buffer (pH 7.4) and was stored at –70 °C until use. At time of experiment, the membranes were diluted in the same ice-cold buffer with 10 μM pargyline (final dilution 1:150, wt/vol). Competition assays were performed in triplicate, in a final volume of 1 mL. To each assay tube were added the following: 0.1 mL of the displacer drug concentration (0.1 mL of vehicle if no competing drug was added) and 0.1 mL of [³H]-raclopride (NEN, 60–87 Ci/mmol) in buffer (final concentration 1 nM). Nonspecific binding was determined using 1 μM cold (+)-butaclamol. Binding experiment was initiated by addition of 0.8 mL of membrane suspension (300–400 μg of protein). After incubation for 60 min at 25 °C the reaction was stopped

for vacuum filtration through Whatman GF/B glass filters, using a Brandel cell harvester, followed by two washes with 5 mL of ice-cold 50 mM Tris-HCl (pH 7.7) buffer. Filters were placed in scintillation poly(ethylene) vials (with 5 mL of scintillation cocktail) and equilibrated, and filter-retained radioactivity was measured in a liquid scintillation counter (Kontron Beta V). IC₅₀ and K_i values were calculated using the computer program EBDA (McPherson).³⁷

4. Rat Isolated Esophageal Tunica Muscularis Mucosae (TMM).^{32,33} Male Wistar rats (250–300 g) were killed by a blow on the head, and a 2-cm segment of intrathoracic esophagus proximal to the diaphragm was removed and placed in standard Krebs-Henseleit (K-H) solution (37 °C, pH 7.4). The external muscularis propria containing the outer longitudinal and circular muscle layers of the esophagus was carefully removed to isolate the inner smooth muscle of TMM. The TMM preparations were suspended longitudinally in a 20-mL tissue bath containing K-H solution at 37 °C, pH 7.4, and continuously aerated with carbogen (95% O₂/5% CO₂). This solution routinely contained 1 μM ketanserin, 10 μM zimelidine, 30 μM cocaine, and 100 μM pargyline. Tissues were placed under an initial tension of 0.5 g, equilibrated for 60 min by washing every 15 min before starting the experiment. Responses were isometrically recorded using a F30 transducer coupled to a two-channel chart recorder (Hugo Sachs Elektronik, Freiburg, Germany). The TMM preparations were contracted by addition of a submaximal concentration of carbachol (3 μM).

In agonism studies, upon establishment of a stable contraction, two concentration–effect curves (relaxation) were constructed in the same TMM in a cumulative fashion: the first for 5-HT itself and the second, 90 min later, for either 5-HT (in time control experiments) or a test agonist. Responses to the cumulative addition of agonist were expressed as percentage relaxation of carbachol-induced tone. For each agonist, two or three separate TMM preparations were used and the concentration–effect curves obtained were fitted to calculate the negative logarithm of the molar concentration of the agonist that relaxed 50% relative to their individual maximum effect (pEC₅₀). The intrinsic activity (i.a.) versus 5-HT was calculated comparing the maximum relaxation obtained at the concentration 1 or 3 μM in the same TMM preparation. To confirm that the agonist effect was 5-HT₄ receptor-dependent, another series with TMM preparations was used in which two concentration–effect curves were constructed: the first to agonist itself and the second, 90 min later, in the presence of 10 nM GR 113808 added 30 min before the first concentration of agonist. Apparent affinity (pK_b) for GR 113808 against all agonists was determined from concentration ratios by the equation: $pK_b = \log(X - 1) + 8$, where X is the concentration ratio and 8 is the negative logarithm of the GR 113808 concentration (10 nM). The effect of this 5-HT₄ antagonist on pEC₅₀ value was registered and then compared by using the unpaired Student's t -test. A p value of 0.05 or less was considered statistically significant.

In antagonist studies, two concentration–effect curves were constructed in the same TMM preparation: the first for 5-HT itself and the second, 90 min later, in the presence of a single concentration (0.1 or 1 μM) of antagonist added 30 min before the first concentration of 5-HT. For each antagonist, two or three separate TMM preparations were used. Apparent affinity (pK_b) for antagonists against 5-HT were determined from concentration ratios by the equation: $pK_b = \log(X - 1) - \log[A]$, where X is the concentration ratio and $[A]$ is the used antagonist concentration (mol/L). The method assumes a competitive interaction. The effect of the antagonists on pEC₅₀ values was registered, and the values were compared by using the unpaired Student's t -test. A p value of 0.05 or less was considered statistically significant.

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Supporting Information Available: Yields, melting points, and spectral data (¹H and ¹³C NMR) for compounds **2–20**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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